

B.Sc. T.Y. (CBCS Pattern) Semester - VI
USDSEPHT13 - Physics Paper-I : Nuclear and Particle Physics

P. Pages : 3

Time : Three Hours



GUG/S/23/13365

Max. Marks : 50

- Notes :
1. All questions are compulsory.
 2. Draw neat labelled diagram wherever necessary.

Either:

1. a) i) Define mass defect and packing fraction. 2
- ii) Explain the concept of Nuclear magnetic dipole moment. What is nuclear magneton? 3
- iii) Discuss electric quadrupole moment of nucleus. 3
- iv) Find the nuclear radius of ${}_{30}\text{Zn}^{64}$ (Given $R_0 = 1.2 \times 10^{-15} \text{ m}$) 2

OR

- b) a) Discuss magnetic moment of an atom. What is Bohr's magneton. 2½
- b) Prove that nuclear density is same for all nucleus. 2½
- c) Discuss various properties of nucleus. 2½
- d) Calculate the ratio of nuclear radius of lead ${}_{82}\text{Pb}^{204}$ and silver isotope ${}_{47}\text{Ag}^{107}$. 2½

Either:

2. a) i) Give the main assumption's of shell model and liquid drop model. 4
- ii) Explain Fermi gas model. 2
- iii) Discuss the concept of nuclear stability. 2
- iv) Explain the concept of nuclear force. 2

OR

- b) a) How does the shell model explains the existence of magic number 2, 8, 20 & 28 only. 2½
- b) Derive the expression for binding energy of nucleus, based on liquid drop model. 2½
- c) Explain the concept of two nucleon separation energy. 2½
- d) Find ground state spin for ${}_{21}\text{Sc}^{45}$ using shell model. 2½

Either:

3. a) i) What is nuclear reaction? Explain various types of nuclear reactions. 3
- ii) Discuss various conservation laws in nuclear reactions. 2
- iii) Derive an expression for Q-value of the reaction $X(a, b)Y$ in terms of kinetic energy. 3
- iv) The Q-value of $\text{Na}^{23}(n, \alpha)\text{F}^{20}$ reaction is -5.4 MeV. Determine the threshold energy of this reaction. 2
- Given : Mass of Neutron = 1.00866 a.m.u
Mass of $\text{Na}^{23} = 22.9909$ a.m.u.

OR

- b) a) Explain the terms Range and Straggling of a charged particle. $2\frac{1}{2}$
- b) Explain the interaction of neutron's with matter. $2\frac{1}{2}$
- c) Discuss the interaction of gamma ray with matter. $2\frac{1}{2}$
- d) Show that the reaction $\text{Li}^7(p, \alpha)\text{He}^4$ is Exothermic. $2\frac{1}{2}$
- Given:
Atomic mass of ${}^1_1\text{H}^1 = 1.00814$ a.m.u., ${}^4_2\text{He}^4 = 4.00260$ a.m.u., ${}^7_3\text{Li}^7 = 7.01822$ a.m.u. .

Either:

4. a) i) Discuss the variation of ionization current with applied voltage due to passage of charged particles through the ionization detector. 3
- ii) Describe construction and working of GM Counter. 3
- iii) Explain working of photomultiplier. 2
- iv) If the frequency of oscillator potential applied to the dees of the cyclotron is 9 MHz. What must be the magnetic flux density to accelerate the α -particles? 2
- Given : Mass of α -particle = 6.643×10^{-27} kg
Charge on α -particle = 3.204×10^{-19} C

OR

- b) a) Explain construction and working of linear accelerator. $2\frac{1}{2}$
- b) Derive the resonance condition of cyclotron? What are limitations of cyclotron? $2\frac{1}{2}$
- c) Describe the working of Van-De-Graff Generator. $2\frac{1}{2}$
- d) What would be the length of last drift tube in a linear accelerator which produces energy 120 MeV C^{12} ions, using frequency of 70 MHz. $2\frac{1}{2}$
- Given : $1\text{eV} = 1.6 \times 10^{-19}$ J, $1\text{amu} = 1.66 \times 10^{-27}$ kg

5. Attempt **any ten** questions from followings carries 1 mark.

- a) Define atomic mass unit. 1
- b) Write the formula for Nuclear Magneton. 1
- c) What is the energy equivalent of 1 a.m.u. 1
- d) Why do the protons in the nucleus not fly apart? 1
- e) What do you mean by charge independence of nuclear force? 1
- f) What are the limitations of liquid drop model? 1
- g) Write Neil's – Bohr's formula. 1
- h) What is Endoergic and Exoergic reaction? 1
- i) Define the term nuclear reaction cross section. 1
- j) State the limitations of linear accelerator. 1
- k) What is mean by threshold voltage in GM tube? 1
- l) Write the principle of ionization chamber. 1
