

B.Sc. S.Y. CBCS Pattern Semester-III
USPHT06 - Physics Paper-II : Radiation and Statistical Physics

P. Pages : 3

Time : Three Hours



GUG/W/23/11617

Max. Marks : 50

-
- Notes : 1. All questions are compulsory.
2. Draw neat labeled diagram wherever necessary.

Either:

1. a) i) State Planck's law for black body radiation. Calculate average energy of Planck's oscillator and derive Planck's radiation from it. 7
- ii) A spherical body has a radius of 0.01 m. this is maintained at 873 K. Calculate the rate at which energy is radiated from its surface assuming it to be a perfectly black body. 3

OR

- b) i) Draw the energy distribution in black body spectrum. What are the conclusions drawn from it. 2½
- ii) State and explain Wien's displacement law. 2½
- iii) A body at 1500 K emits maximum energy at wavelength $20,000 \text{ \AA}$. If the Sun emits maximum energy at wavelength at 6666 \AA , what would be the temperature of the Sun? 2½
- iv) Derive Stefan's Boltzmann law from Planck's radiation law. 2½

Either:

2. a) i) Explain the concept of phase space and μ – space. 3
- ii) Discuss the constraints in thermodynamic system. 2
- iii) What is most probable macrostate? 2
- iv) In a system, 8 distinguishable particles are distributed in two equal sized compartments. Calculate the probability of the macrostate (3, 5), (4, 4) and (2, 6) 3

OR

- b) i) Explain the statistical meaning of entropy. 2½
- ii) Explain the term principle of equal a priori probability. 2½
- iii) Calculate the number of accessible microstate W of a system having entropy 10 cal K^{-1} in the equilibrium state. 2½

iv) Define thermodynamic probability and state its minimum value. 2½

Either:

3. a) i) Using M. B. distribution law, $n_i = g_i e^{-\alpha} e^{-(E_i/kT)}$ derive an expression for Maxwell Boltzmann energy distribution law for the particles of an ideal gas, By evaluating the multiplier α . 8

ii) Calculate the most probable velocity of a hydrogen at 27° C. 2
Given: Boltzmann's constant $k = 1.38 \times 10^{-23} \text{ J / deg}$ and
Mass of hydrogen molecule = $3.34 \times 10^{-27} \text{ kg}$

OR

b) i) State the characteristics of a particle obeying M. B. Statistics. 2½

ii) Using Maxwell's law of distribution of speeds of molecules in a gas, obtain an expression for most probable speed. 2½

iii) At what temperature will the average speed of molecules of hydrogen gas be double the average speed of oxygen at 300 K? 2½

iv) Derive an expression for root mean square speed of a molecules of ideal gas. 2½

Either:

4. a) i) Derive an expression for Fermi Dirac distribution law for particles obeying F. D. statistics. 6

ii) Derive an expression for B. E. energy distribution law for continuous variation of energy from B. E. distribution. 2

iii) Five bosons are distributed in two compartments. The first having 3 cells and the second 4. Find thermodynamic probability for the microstate (5, 0). 2

OR

b) i) Write the fundamental postulates of Bose-Einstein statistics. 2½

ii) Differentiate between M. B. and F. D. statistics. 2½

iii) Show that the Fermi energy of electron in metal at absolute zero is 2½
 $E_{F(0)} = (3n / 8\pi)^{2/3} h^2 / 2m$ Where, the symbols have their usual meanings.

iv) Calculate the number of different arrangements of 10 indistinguishable particles in 15 cells of equal a priori probability considering that one cell contains only one particle. 2½

5. Solve **any ten** from the following.

a) Define Black body. 1

- b) State Planck's radiation law in terms of frequency. 1
- c) State Stefan-Boltzmann law. 1
- d) Define most probable distribution. 1
- e) Two coins are tossed simultaneously. Find the probability of getting one head and two tails. 1
- f) Define statistical weight. 1
- g) Give the limitations of M. B. statistics. 1
- h) The speed of ten particle in m/sec is 0, 1, 2, 3, 3, 4, 4, 4, 5 and 6 respectively. What is their most probable speed? 1
- i) Define average speed of gas molecules. 1
- j) What are Fermions. 1
- k) What is occupation index? 1
- l) Give the postulates of Fermi-Dirac statistics. 1
